

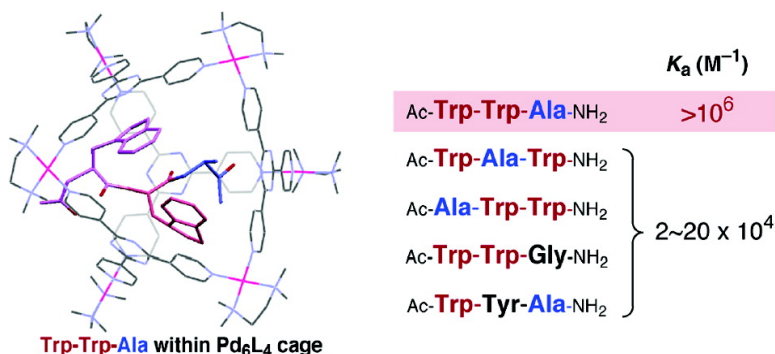
Communication

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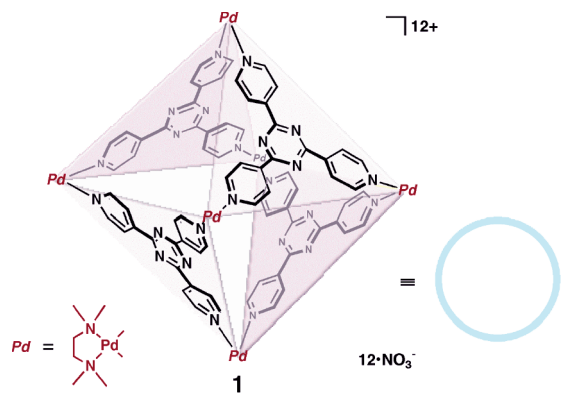
Sequence-Selective Recognition of Peptides within the Single Binding Pocket of a Self-Assembled Coordination Cage

Shohei Tashiro,[†] Masahide Tominaga,[†] Masaki Kawano,[†] Bruno Therrien,[†] Tomoji Ozeki,[‡] and Makoto Fujita^{*†}

Department of Applied Chemistry, School of Engineering, The University of Tokyo, CREST, Japan Science and Technology Agency (JST), 7-3-1 Hongo, Bunkyo-ku, Tokyo 113-8656, Japan, and Department of Chemistry and Materials Science, Tokyo Institute of Technology, 2-12-1 O-okayama, Meguro-ku, Tokyo 152-8551, Japan

Received August 30, 2004; E-mail: mfujita@appchem.t.u-tokyo.ac.jp

Sequence-selective recognition of peptides is expected to be an essential process for the site-specific recognition of protein surfaces,¹ which leads to the control of protein functions and to the understanding of biological events at protein surfaces such as protein–protein or protein–oligopeptide interactions.² Although a few groups have reported artificial receptors for sequence-selective peptide recognition, the highly designed receptors are in their early stages. They are constructed by simply linking small recognition sites such as cyclodextrins,³ crown ethers,⁴ or amide groups,⁵ which all bind only a single amino acid residue. Here, we report that the single binding pocket of self-assembled coordination cage **1**⁶ can accommodate oligopeptides in a highly sequence-selective fashion. Having a large hydrophobic cavity, cage **1** binds as many as three amino acid residues. NMR and X-ray analyses reveal that the sequence-selective recognition is ascribed to cooperative multiple interactions between the residues and the cavity.



In a first series, the sequence-selective recognition of tripeptides, Ac-X¹-X²-X³-NH₂ (X^{1–3} = amino acid residues), was examined.⁷ We found that cage **1** bound Ac-Trp-Trp-Ala-NH₂ (**2**) very strongly ($K_a > 10^6 \text{ M}^{-1}$).⁸ The **1**·**2** complex was easily prepared by suspending powdered **2** (1 mol equiv) in a D₂O solution of **1** (10 mM) at room temperature for 0.5 h. Strong binding was specific to the Trp-Trp-Ala sequence because the binding of tripeptides possessing those same residues in different sequences, such as Ac-Trp-Ala-Trp-NH₂ (**3**) and Ac-Ala-Trp-Trp-NH₂ (**4**), was much less effective ($K_a = 2.5 \times 10^5$ and $2.1 \times 10^4 \text{ M}^{-1}$, respectively). Even singly mutated tripeptides, such as Ac-Trp-Trp-Gly-NH₂ (**5**) and Ac-Trp-Tyr-Ala-NH₂ (**6**), showed poorer affinity ($K_a = 7.4 \times 10^4$ and $5.3 \times 10^4 \text{ M}^{-1}$, respectively) although they have very similar aromatic–aromatic–aliphatic sequences (Table 1). These results

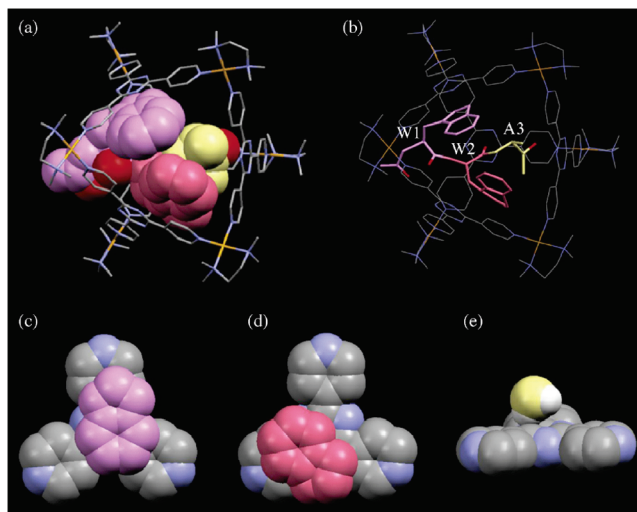


Figure 1. Crystal structure of **1**·**2**. Peptide **2** in the cavity is represented by (a) space-filling and (b) cylindrical model. The π – π interactions of **1** with indole rings of (c) W1 and (d) W2. (e) The CH– π interaction between **1** and methyl group of A3.

Table 1. Association Constants of **1** with Peptides in Water

Peptides	$K_a \text{ (M}^{-1}\text{)}^a$
Ac-Trp-Trp-Ala-NH ₂ (2)	$>10^6$
Ac-Trp-Ala-Trp-NH ₂ (3)	2.5×10^5
Ac-Ala-Trp-Trp-NH ₂ (4)	2.1×10^4
Ac-Trp-Trp-Gly-NH ₂ (5)	7.4×10^4
Ac-Trp-Tyr-Ala-NH ₂ (6)	5.3×10^4
Ac-Gly-Gly-Ala-NH ₂ (7)	no binding
Ac-Trp-His-Ala-NH ₂ (8)	no binding
Ac-Tyr-Tyr-Ala-NH ₂ (9)	4.7×10^3
Ac-Ala-Trp-Trp-Ala-NH ₂ (10)	$>10^6$
Ac-Ser-Gly-Ala-Trp-Trp-Ala-NH ₂ (11)	$>10^6$

^a Measured by UV–vis titration at 20 °C.

suggest that the two indole rings and the Ala methyl group in **2** should cooperatively interact with the cage in the **1**·**2** complex.

In fact, the multiple interactions of the methyl and indole groups with the cage were revealed by X-ray crystallographic analysis.⁹ Single crystals were obtained after an aqueous solution of **1**·**2** complex stood at room temperature for 4 d. The diffraction data were collected by synchrotron X-ray irradiation. The crystallographic analysis showed that tripeptide **2** is fully encapsulated in the cavity of **1** (Figure 1a,b). As predicted, all residues interact very efficiently with cage **1**. Namely, two indole rings are stacked on the triazine ligand by π – π interaction (3.4–3.5 Å), while the Ala methyl group interacts with another ligand by CH– π contact

[†] The University of Tokyo, CREST, Japan Science and Technology Agency (JST).

[‡] Department of Chemistry and Materials Science, Tokyo Institute of Technology.

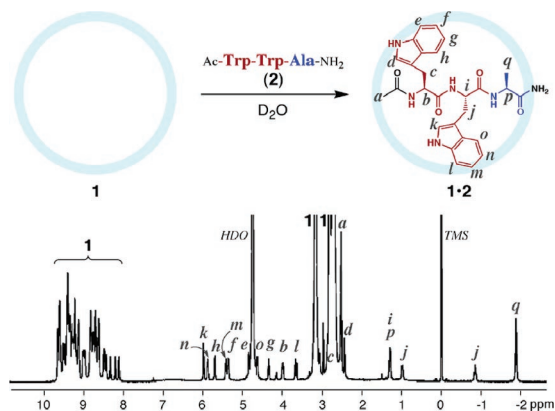


Figure 2. ^1H NMR of **1·2** in D_2O (500 MHz, 10 mM, 27°C , TMS as external standard).

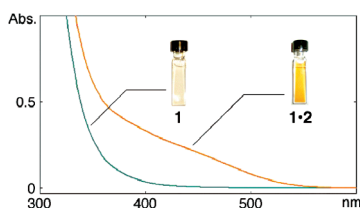


Figure 3. UV-vis spectra of free cage **1** and complex **1·2** (H_2O , 0.2 mM, rt).

(2.5 Å) (Figure 1c–e). Despite the enclathration within the restricted cavity, the peptide backbone is fixed in an extended conformation.

The inclusion geometry shown by the X-ray analysis is in good agreement with the NMR observations. In the ^1H NMR, the Ala methyl signal at $\delta -2.0$ and the indole aromatic protons around $\delta 6.0$ – 2.5 are considerably upfield shifted due to the shielding effect of the cage (Figure 2). A clear NOE correlation between the Ala methyl group (signal *q* in Figure 2) and one indole ring (signal *e*) was observed, which was explained by the tight contact of these proton pairs (2.6 Å) as revealed by X-ray analysis. Similarly, two indole rings (signals *g* and *n*) are correlated by NOE and shown to be in close contact (2.8 Å). In the $\delta 10.0$ – 8.0 region, the pyridyl protons of the cage were observed in a very complex pattern indicating the desymmetrization of the cage. The motion of **2** is restricted by enclathration, and therefore all pyridine protons of the cage become inequivalent. We note that the clear desymmetrization of the cage in NMR is only observed for **2**. It seems that the motions of other tripeptides in the cavity are not strictly restricted, and therefore, the ^1H NMR cage signals are simply broadened.

The efficient π – π stacking observed by X-ray and NMR is ascribed to charge transfer from the indole rings to the electron-deficient triazine ligand.¹⁰ The color of the solution turned yellow upon formation of the **1·2** complex. In the UV-vis spectrum, broad absorption around 350–550 nm was observed (Figure 3).

The sequence selectivity changes significantly when electronically and sterically different tripeptides are used. For instance, Ac-Gly-Gly-Ala-NH₂ (**7**) that possesses no aromatic residue was not bound at all. Ac-Trp-His-Ala-NH₂ (**8**) that involves a cationic residue (His) did not show any interactions with the cage due probably to cationic repulsion with the cage. Ac-Tyr-Tyr-Ala-NH₂ (**9**) was bound $>10^2$ times more weakly than **2**, indicating the importance of electron donation from electron-rich indole rings to the electron-deficient ligand¹⁰ (Table 1).

Accordingly, cage **1** recognizes specifically the Trp-Trp-Ala sequence in oligopeptides consisting of more than three amino acid

residues. For example, tetrapeptide **10** and hexapeptide **11**, both possessing the Trp-Trp-Ala sequence, were strongly bound to **1** ($K_a > 10^6 \text{ M}^{-1}$, respectively).⁸ This specificity is confirmed by NMR where only the Trp-Trp-Ala part of these oligopeptides was dramatically upfield shifted.¹¹

In summary, we have shown the sequence-selective recognition of peptides by the single binding pocket of cage **1**. As organic modification of the cage is easy and related large, hollow structures with different shapes and sizes have been previously prepared,¹² the design of single-pocket receptors for specific sequences of oligopeptides is our next challenge. Ultimately, the pinpoint recognition of protein surfaces by these self-assembled hollow receptors is the main goal of the present study.

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Supporting Information Available: Experimental procedures, physical properties of **1·2**, NMR spectra, UV-vis titrations (PDF) and crystallographic data of **1·2** (CIF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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- (7) Association constants were measured by the UV titration at a charge-transfer band in water. Spectra were analyzed by a nonlinear curve-fitting procedure. See the Supporting Information.
- (8) Since UV titrations of **1·2**, **1·10**, **1·11** showed the saturation at nearly 1 equiv of **1**, exact values of K_a could not be calculated by this measurement. Thus, their K_a 's were estimated to be 10^6 M^{-1} .
- (9) Crystal data for **1·2**: $\text{C}_{135}\text{H}_{202}\text{N}_{54}\text{O}_{54}\text{Pd}_6$, $M = 4083.90$, Monoclinic, space group $C2/c$, cell parameters $a = 70.38(1) \text{ \AA}$, $b = 17.181(3) \text{ \AA}$, $c = 42.159(8) \text{ \AA}$, $\beta = 99.90(3)^\circ$, $V = 50217(2) \text{ \AA}^3$, $T = 15(2) \text{ K}$, $Z = 8$, $D_c = 1.068 \text{ g cm}^{-3}$, λ (synchrotron) = 0.6890 \AA , 142086 reflections measured, 47250 unique ($R_{\text{int}} = 0.0714$) which were used in all calculations. The structure was solved by direct method (SHELXL-97) and refined by full-matrix least-squares methods on F^2 with 2114 parameters. $R1 = 0.1518$ ($I > 2\sigma(I)$) and $wR2 = 0.4521$, $\text{GOF} = 1.378$; max/min residual density $1.532/-1.824 \text{ e \AA}^{-3}$. CCDC reference number 247331. It is noteworthy that in the crystal of **1·2**, the 12 nitrate anions as well as the water molecules were highly disordered, giving rise to a large R1 value. However, the guest molecule was found from Fourier difference maps, and only minor disorder over the peptide backbone chain was found.
- (10) Four triazine ligands composing cage **1** are highly electron deficient due to multiple electron withdrawing from three palladium cations.
- (11) For the NMR assignments of a site-specific recognition of **1·10** and **1·11**, see the Supporting Information.
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